TRACKING LIPID OXIDATION IN STORED SHELLED AND INSHELL RAW ALMONDS BY HS-SPME-GC

California almonds

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HS-SPME-GC-FID and -MS were used to track the extent of lipid oxidation in shelled and inshell raw almonds, stored under different temperature/relative humidity (RH) conditions, over 24 months. In total, 5 aldehydes, 13 alcohols, 4 alkanes, and 10 miscellaneous organic volatiles were identified and quantified. The extraction temperature, extraction time, and desorption period of major almond volatiles were optimized. Quantitation was carried out in two ways: calibration curves of hexanol-d₁₃ and hexanal-d₁₂, relating peak area to concentration, were developed; and the standard addition method was performed using these isotopologic-deuterated standards and an ultrapycnometer to determine the density of ground almond powder (GAP). With both methods of quantitation yielding <12% differences in calculated µg/mL concentrations of volatiles, the employment of deuterated isotopologues of analytes-of-interest was validated. Relative quantitation of these volatiles indicated that storage parameters (i.e., temperature and % RH) were influential in their development, especially for volatiles that were known secondary lipid oxidation products of linoleic acid. To illustrate, hexanal levels within the first 8 mo of storage at 25 °C achieved levels of 5.5 and 7.2 µg/g GAP when stored under low- and high-humidity conditions, respectively. Hence, the comparison of volatile contents between inshell and shelled almonds throughout storage highlights a potential for the utilization of inshell storage as a means of promoting flavor stability, although this efficacy may be dependent on shell characteristics of the almond cultivar in question.

Materials & Methods

Raw shelled and inshell almonds (varieties: 'Nonpareil' & 'Butte') were stored in woven PP bags at varying temperature and % RH conditions in SP Scientific environmental chambers for up to 24 mo.

Ground Almond Powder (GAP) for GC Analyses

- $\circ\,$ At 8-mo intervals, almonds were randomly sampled, shelled using a nut cracker (for inshell samples), immersed in liquid N_2, ground, and then sieved using an 18-mesh Tyler standard screen to give a ground almond powder (GAP).
- GAP samples (2.5 g) were transferred into 20-mL amber glass HS vials, sealed with an AI HS cap and PTFE/silicone liner, and stored at -80°C until GC analysis.

Volatile extraction and HS-SPME Conditions

- $_{0}$ To GAP samples, 1.7 mL of a saturated NaCl solution, 80 μ L of hexanol-d₁₃, and 90 μ L of hexanal-d₁₂ (internal standards both at 40 μ g/mL) were added; contents were vortexed for 60 s.
- CTC Combi PAL autosampler set at 55°C for 40 min using a 50/30 μm DVB/CAR/PDMS SPME fiber was employed to adsorb volatiles.
 GC-FID and –MS Conditions
- Agilent 7890A GC with FID at 275°C; Combi PAL autosampler; J&W DB-5 column (60 m x 0.25 mm; 0.25 µm); inlet at 250°C; split ratio 5:1; 8 min desorption from SPME fiber; 1 mL/min He flowrate; oven from 38°C to 175°C at 2.5°C/min, then to 250°C at 20°C/min.

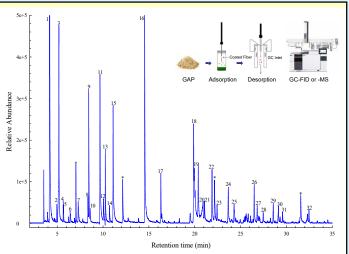
Same setup with MS detector using NIST mass spectral library. Quantitation of Volatiles from Calibration Curves of the Iss

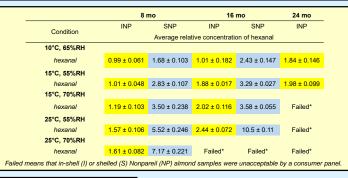
- Standard addition method with hexanol-d₁₃ and hexanal-d₁₂ was used to quantitate endogenous levels of hexanol and hexanal in the GAP samples.
- A Quantachrome ultrapycnometer 1000 was used to determine the averaged density (g/cm³) of prepared GAP samples by using argon gas to employ Archimedes' principle of fluid displacement.

Results & Discussion

- From the Tables & Figure: 5 aldehydes, 13 alcohols, 4 alkanes, and 10 miscellaneous organic HS volatiles were identified for shelled and in-shell almonds, and quantified at 8, 16, and 24 mo of storage.
- Greatest change in HS levels was for hexanal: its concentrations at 8, 16, and 24 mo (unless the sample failed sensorially) are given.
- $_{\rm O}\,$ Calibration curves for hexanol-d_{13} and hexanal-d_{12} in devolatilized GAP samples had R² values of 0.992 and 0.982, respectively.
- Using an ultrapycnometer and the 'standard addition method' with the deuterated standards, HS volatile concentrations were different by <12% relative to values calculated from the calibration curves.
- As temp/%RH conditions became more abusive (e.g., 25°C, 70% RH vs 15°C, 70% RH), hexanal levels were found to be greater.
- The shell of 'Nonpareil' almonds (INP) afforded significantly (p<0.05) more protection against secondary lipid oxidation volatile development compared to shelled (SNP) counterpart samples.







Acknowledgements

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Odor Description

fermented, nutty, fruity, berry

green, grassy, leafy, sweaty

sweet, bitter, almond, cherry

aldehydic, citrus, green, fatty

aldehydic, rose, peely, waxy

oily, sweet, balsamic, whiskey

fusel, whiskey, fruity, banana

fusel, oily, sweet, balsamic

fruity, green, lavender, yeasty

herbal, fruity, alcoholic, sweet

floral, rose, phenolic, balsamic

fresh, floral, rose, orange, oily

floral, rose, dried rose

gasoline-like odorless when pure, gasoline

gasoline-like

gasoline-like

aromatic

camphoraceous

sour, fatty, sweaty, cheesy

ethereal, apple, pear, solvent

ethereal, winey

sweet, fruity

ethereal, fresh

creamy

Tentative identifications of aroma-active cmpds in raw almonds after 24-mo

Analyte

ntanal

exanal

octanal

onanal

1-butanol

1-pentanol

2,3-butanediol

benzyl alcohol

henylethyl alcoho

entane, 3-methyl

cyclopentane, methyl

xime-, methoxy-phenyl-_

enzene, 1,2,3-trimethyl-

nzene, 1.2.3.4-tetramet

enzene, 1-ethyl,2,4-dimethyl- plastic

nzene, 1-ethyl-2-methy

1-hexanol

-nonanol

7-hexane

clooctane

xanoic acid

sopropylbenzene

etone

somers

orenol

I-propanol, 2-methyl

3-methylbut-3-en-1-ol

1-butanol, 2-methyl-, (S)

2,3-butanediol, {R-(R*,R*)]-

3-methyl-1-butanol

enzaldeh

Numbering

Aldehydes

7

15

18

24

29

Alcohols

4

6

8 9

10

11

12

13 14

16

26

30

32

<u>Alkanes</u> 2

3

27

Misc. cmpds

21

17

19

20

22, 23, 25

28

31