

INFLUENCE OF MATRIX COMPOSITION ON THE EFFECT OF HYDROXYTYROSOL ACETATE ON OIL OXIDATION

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INTRODUCTION

The oxidative stability of oils rich in polyunsaturated acyl groups is a crucial aspect that influences their quality, since oxidation can lead to nutrient losses, development of off-flavours and generation of toxic compounds. Consequently, much research has been carried out on the effect of numerous compounds and extracts with potential antioxidant capacity on edible oil stability (1). In this context, hydroxytyrosol acetate has been shown to be effective in reducing oil oxidation under accelerated storage conditions (2). Notwithstanding, more information is needed in order to be able to select the most appropriate antioxidants for different types of oils, since their composition could influence the observed effect.

OBJECTIVE

To study the influence of matrix composition on the effect of hydroxytyrosol acetate on the oxidation process of three different types of edible oils, all of them very rich in polyunsaturated acyl groups.

MATERIALS AND METHODS

SAMPLES: Refined sunflower oil (RSO), virgin linseed oil (VLO) and refined microalgae oil containing 5000 ppm of an extract rich in tocopherols (E-306) (RMO), and these same oils enriched with hydroxytyrosol acetate (OHTyrAc) in a proportion of 0.1% by weight: RSO + OHTyrAc, VLO + OHTyrAc and RMO + OHTyrAc.

OXIDATION: The oils (6.5 g in glass Petri plates of 5 cm diameter) were submitted to an accelerated storage process at 70 °C in an oven with aeration. For each oil, aliquots were taken periodically from the same plate until total polymerization.

STUDY OF THE SAMPLES: The samples were studied before and at different times of the accelerated storage process by ¹H Nuclear Magnetic Resonance (¹H NMR).

RESULTS AND DISCUSSION

1. SOME COMPOSITIONAL ASPECTS OF THE STUDIED OILS

As Table 1 shows, the three types of oils differ both in their acyl group composition and in some minor components with attributed antioxidant ability, such as tocopherols. Differences in acyl groups are reflected in the unsaturation degree of the oils ($R_{S/U}$), RMO being the most unsaturated one and RSO the least. While in RSO α -tocopherol (α -T), the main tocol present in this type of oil (3), was not in high enough concentration to be detected by ¹H NMR, in RMO an elevated level of γ -T was found.

Table 1. Mean values of molar percentage of the main polyunsaturated (PU) group present in each oil, degree of unsaturation ($R_{S/U}$), the most abundant tocopherol (T) in each oil and its concentration, expressed in millimoles per mole of triglyceride (mmol/mol TG).

	RSO	VLO	RMO
Main PU group / molar %	Linoleic / 60.1	Linolenic / 47.2	ω -3* / 43.5
$R_{S/U}$ (ratio of saturated to unsaturated protons)	9.9	6.4	2.5
Main tocopherol / mmol/mol TG	α -T / nd	γ -T / 1.10	γ -T / 9.85

*: mainly DHA; nd: not detected

2. EVOLUTION OF THE CONCENTRATION OF THE MAIN POLYUNSATURATED ACYL GROUPS AND OF OHTyrAc

- A reduction in the degradation rate of the samples enriched with OHTyrAc is observed (see Figure 1), which leads to an enlargement in the time needed to reach oil total polymerization. The extent of this effect is considerably pronounced for RSO sample, in accordance with previous studies (2), less marked for VLO and very slight for RMO. Consequently, after 9 days, the greatest decrease in the loss of the main polyunsaturated acyl group is found in RSO + OHTyrAc, the least unsaturated oil (see Table 1), whereas differences are hardly noticed in RMO, the most unsaturated one.
- The concentration of OHTyrAc decreases with time (see Figure 2), although the rate of this drop varies depending on the oil. Thus, OHTyrAc takes a little longer to degrade in RSO, where the antioxidant effect of this compound seems to be the most intense (see Figure 1), than in VLO. Instead, in RMO oil OHTyrAc diminishes much more rapidly. Considering previous results obtained with mixtures of OHTyrAc and α -T (4), it could be thought that, similarly to α -T, γ -T, present in high concentration in RMO (see Table 1), might have interacted with OHTyrAc, counteracting to some extent its antioxidant potential in this oil.

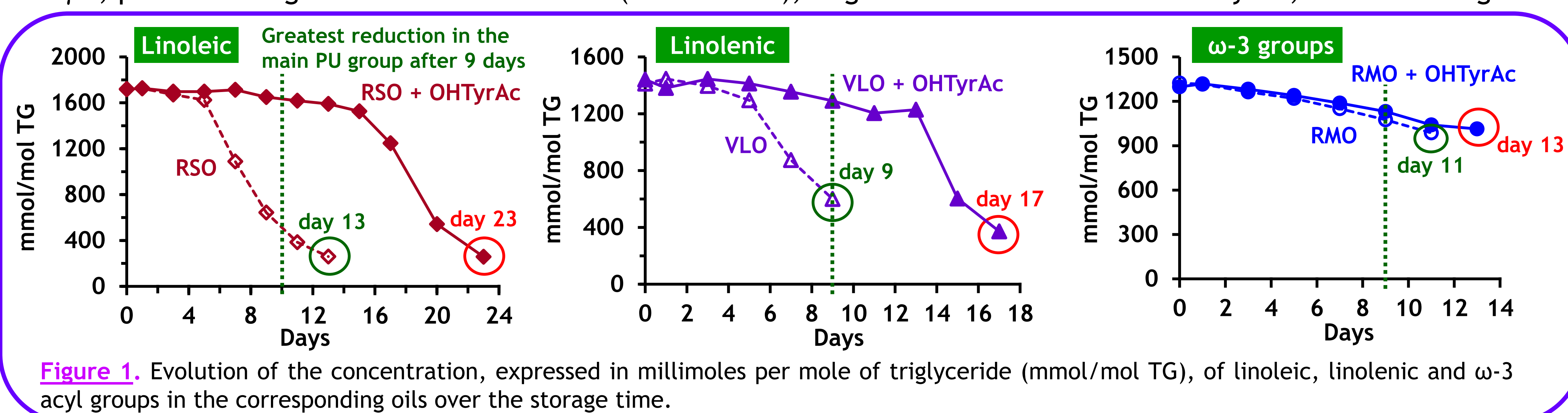


Figure 1. Evolution of the concentration, expressed in millimoles per mole of triglyceride (mmol/mol TG), of linoleic, linolenic and ω -3 acyl groups in the corresponding oils over the storage time.

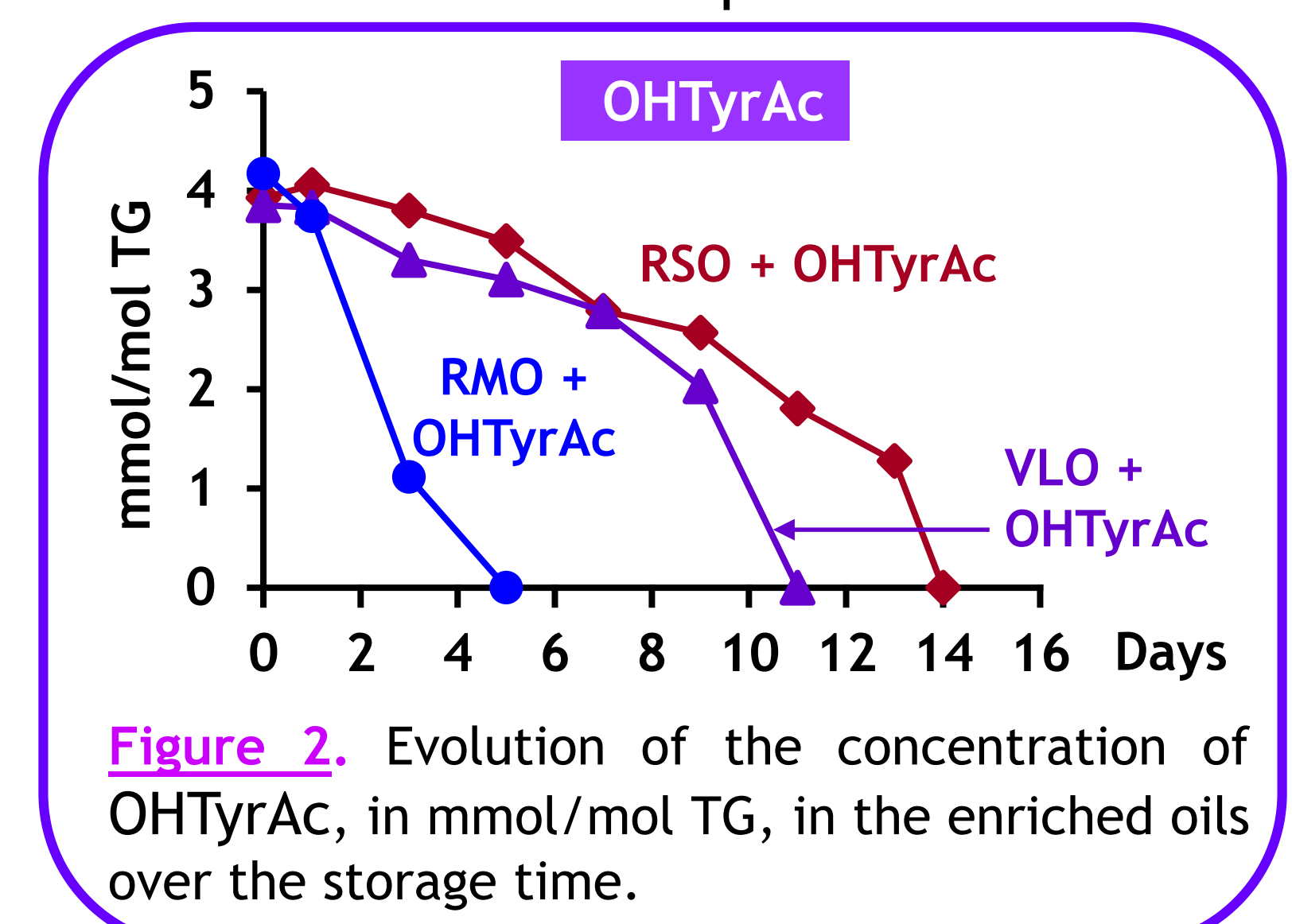


Figure 2. Evolution of the concentration of OHTyrAc, in mmol/mol TG, in the enriched oils over the storage time.

3. GENERATION OF OXIDATION PRODUCTS

- The effect of the OHTyrAc enrichment on the formation of oxidation derivatives is in accordance with that observed on the evolution of the main polyunsaturated acyl group in each oil. Thus, considerable reductions in the generation rate of hydroperoxides (data not shown) were noticed especially in RSO, but also in VLO. However, in RMO the addition of OHTyrAc hardly prevents the formation of oxidation compounds.
- The delay in primary oxidation compound generation results in a later appearance of those secondary ones like aldehydes. This deserves special interest, since the addition of OHTyrAc greatly reduces the formation of some potentially toxic oxygenated α,β -unsaturated ones in RSO and VLO during a considerable period of time (see Figure 3).

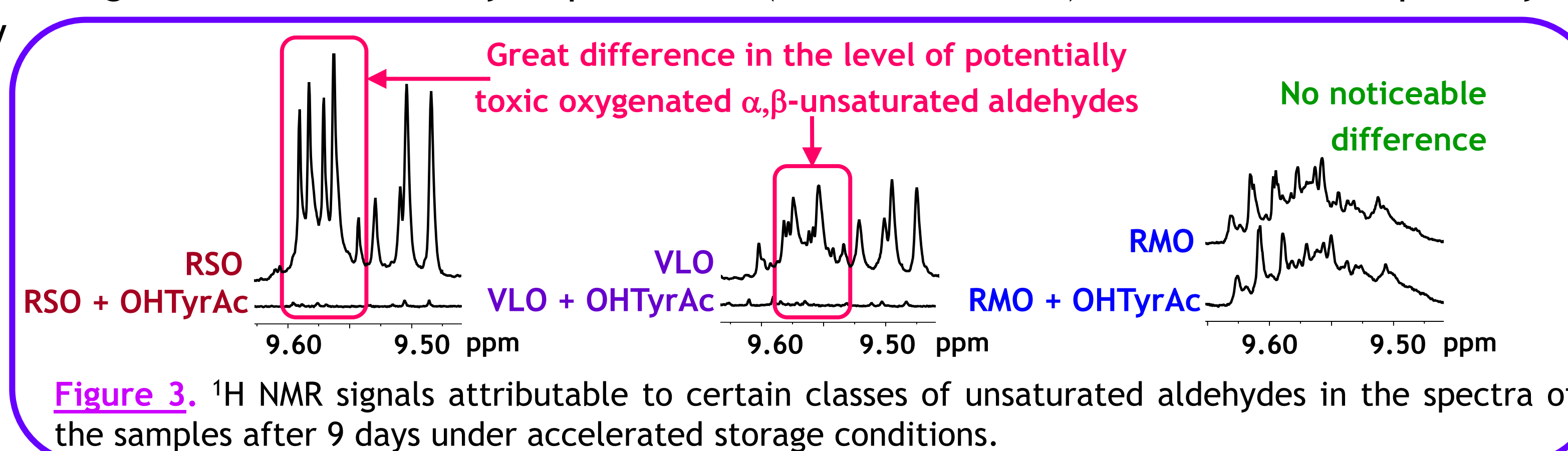


Figure 3. ¹H NMR signals attributable to certain classes of unsaturated aldehydes in the spectra of the samples after 9 days under accelerated storage conditions.

CONCLUSIONS

- The effect of OHTyrAc on oil oxidation rate is influenced by the matrix composition, it being very pronounced for sunflower, and to a somewhat lesser extent, for linseed oil, but very weak for microalgae oil. These findings suggest that the higher the unsaturation degree of the oils, the weaker the antioxidant effect of OHTyrAc. However, the high concentration of γ -T in the microalgae oil might also have contributed to the scarce effect of OHTyrAc observed in this oil compared with the other ones.
- Further studies are needed in order to know the possible interactions between OHTyrAc and other minor components that can be present in edible oils, either naturally or because of an addition, in order to elucidate how these could affect its behaviour.

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